Trends in California Gasoline Properties and Motor Vehicle Emissions

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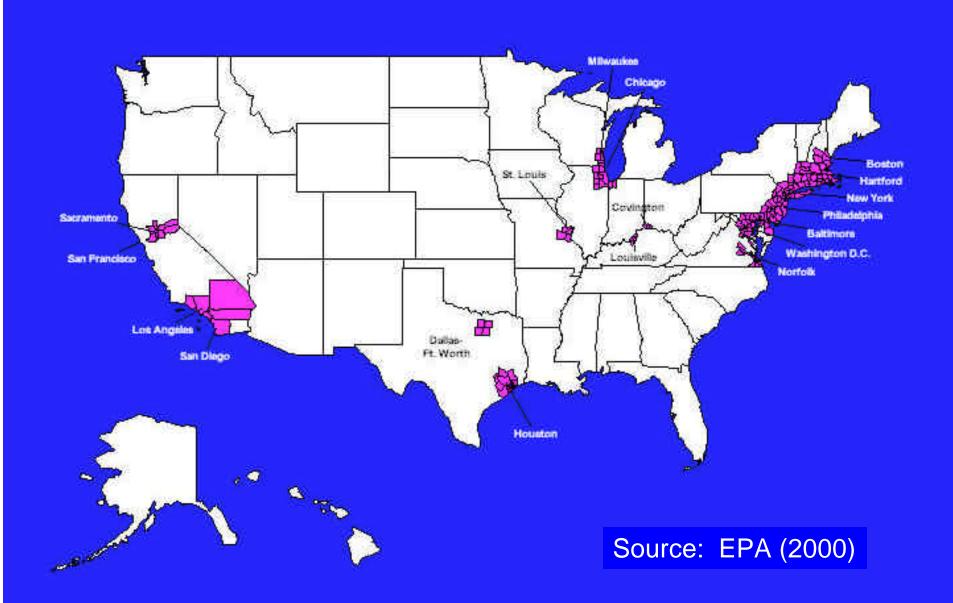
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 - California Air Resources Board

Introduction

- Major changes to gasoline since 1990:
 - Wintertime use of oxygenates (1992)
 - California RFG (1992, 1996)
 - Federal RFG (1995, 2000)
- Further changes are underway!
 - MTBE will be phased out in California by end of 2002

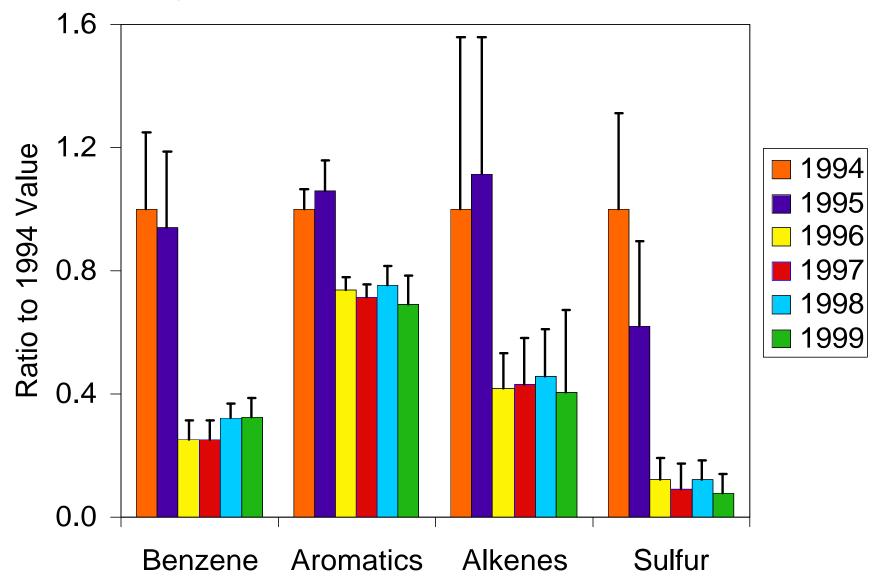
Federal Reformulated Gasoline Areas



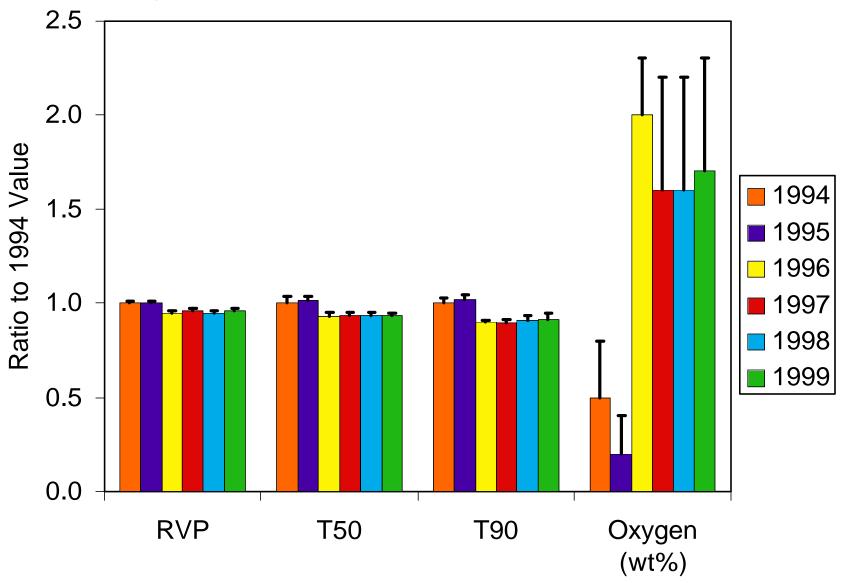
Bay Area Gasoline (1994)

Benzene (vol%)	1.6 <u>+</u> 0.4	RVP (psi)	7.4 <u>+</u> 0.1
Aromatics (vol%)	31.9 <u>+</u> 2.1	T ₅₀ (°F)	214 <u>+</u> 8
Alkenes (vol%)	7.9 <u>+</u> 4.4	T ₉₀ (°F)	334 <u>+</u> 8
Sulfur (ppmw)	131 <u>+</u> 41	Oxygen (wt%)	0.5 <u>+</u> 0.3

Bay Area Gasoline Trends



Bay Area Gasoline Trends



Vehicle Emissions

- Most studies of fuel effects have relied on laboratory dynamometer testing
 - Test one vehicle at a time
 - Simulate stop-and-go city driving
 - Repeat using different fuel formulations
- For example, see Auto/Oil study (Hochhauser et al., SAE 912322)

Caldecott Tunnel

- Tunnel on hwy. 24 east of Oakland, CA
 - 1100 meters (0.7 mile) long
 - Three two-lane traffic tubes
 - Sample in middle bore (no diesel trucks)
 - Sample 10 days each summer 1994-99
 - Sample 4-6 PM (>4000 vehicles/hour)
 - Traffic is eastbound/uphill on 4.2% grade



Pollutant Measurements

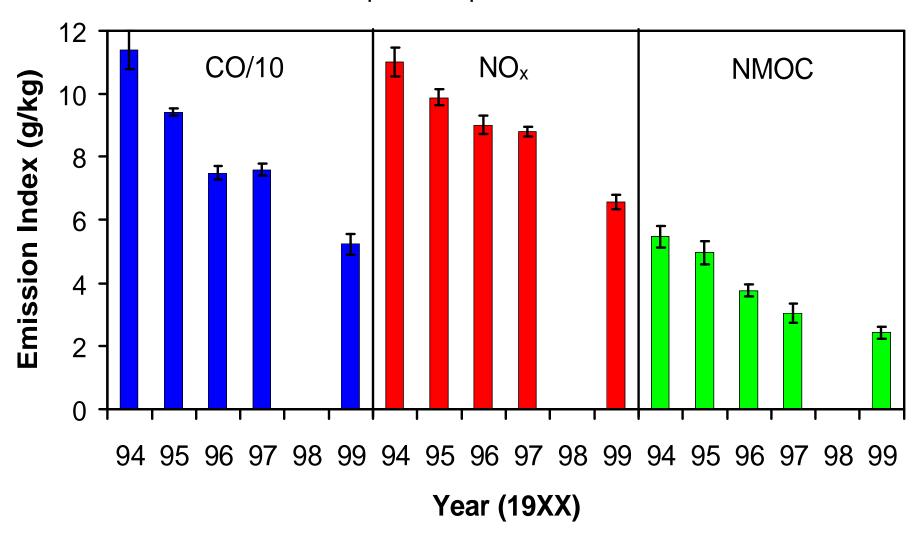
- Carbon dioxide (CO₂)
- Carbon monoxide (CO)
- Nitrogen oxides (NO_x)
- Non-methane hydrocarbons (NMHC)
- Methyl tert-butyl ether (MTBE)
- Methane (CH₄)
- Aldehydes

Pollutant Measurements

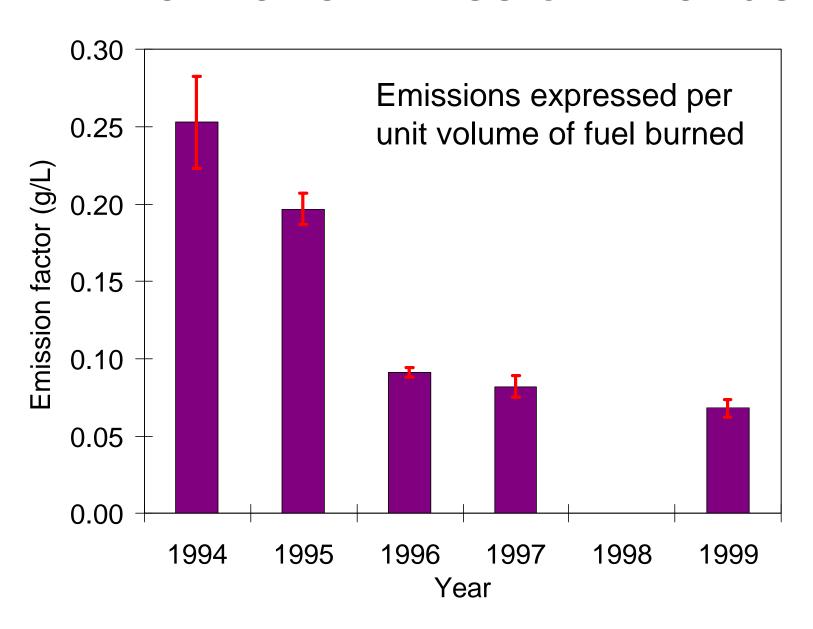


Emission Factor Trends

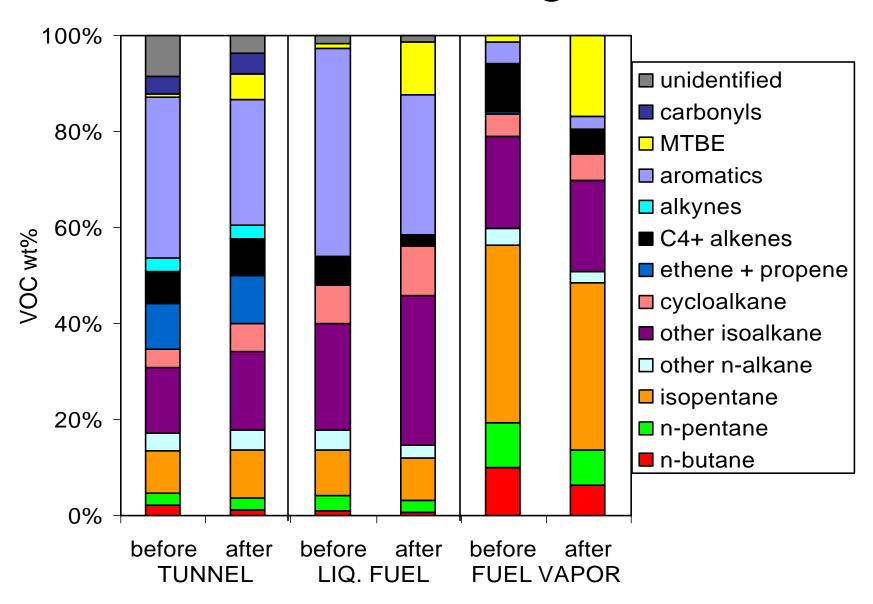
Emissions expressed per unit mass of fuel burned



Benzene Emission Trends



Effects of Fuel Change on VOC



Summary

- Emission factors decreased between 1994 and 1999;
 - Benzene down by 73 <u>+</u>12%
 - NMOC down by 55 ± 7%
 - CO down by 54 ± 6%
 - NO_x down by 41 ± 4%
- Improved vehicle technology more important than fuel changes, except benzene where contribution due to fuel changes is 30-40%

Air Quality Issues: Ethanol

 Acetaldehyde (CH₃CHO) emissions will increase by ~150% if ethanol added to gasoline at 10 vol% (SAE 920326)

CH₃CHO + OH'
$$\xrightarrow{O_2}$$
 CH₃C(O)O'₂ + NO₂

$$\updownarrow$$
CH₃C(O)OONO₂

Peroxyacetyl nitrate (PAN)

Vapor Pressure

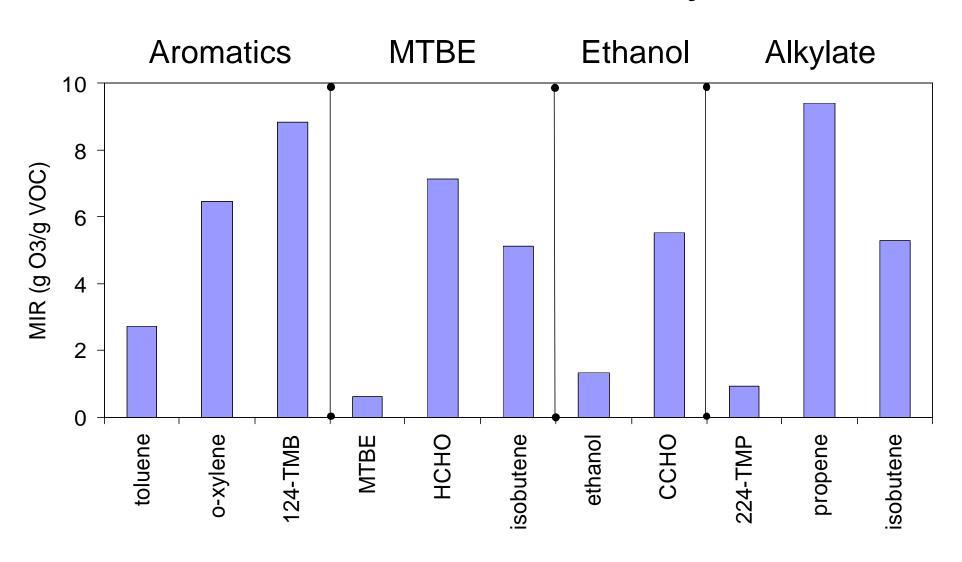
- Affects tendency of gasoline to evaporate in hot weather
- Gasoline v.p. limit 48 kPa at 38°C
- Ethanol has v.p. of 16 kPa at 38°C
- Adding ethanol should lower v.p. of gasoline... right?

No! It forms non-ideal solution and vapor pressure increases.

Air Quality Issues: Alkylate

- Strong acid catalysts (HF, H₂SO₄) used in refinery alkylation process are hazardous
- Alkylate in gasoline is a precursor to emissions of C₃-C₄ alkenes (highly reactive) in vehicle exhaust
- Higher energy content than oxy-fuels

VOC Reactivity



Conclusions

- Reactivity of exhaust VOC with respect to ozone formation is unlikely to change for MTBE vs. ethanol vs. alkylate
- Potential air quality impacts:
 - Vapor pressure problems, increased acetaldehyde & PAN formation (for EtOH)
 - Hazardous strong acid catalysts used in refineries to make alkylate

References: Environ. Sci. Technol.

- Kirchstetter et al., vol. 33, pp. 318-28, 1999a.
- Kirchstetter et al., vol. 33, pp. 329-36, 1999b.
- Kean et al., vol. 34, pp. 3535-39, 2000.
- Harley et al., vol. 34, pp. 4088-94, 2000.